Cryopreservation of Structural Integrity under High Pressure

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Cornell High Energy Synchrotron Source

Science at the Hard X-ray Diffraction Limit June 6, 2011



Cornell University MacCHESS Macromolecular Diffraction at Cornell High Energy Synchrotron Source



High Pressure Cryocooling & X-ray Diffraction Microscopy (XDM) of Biological Samples

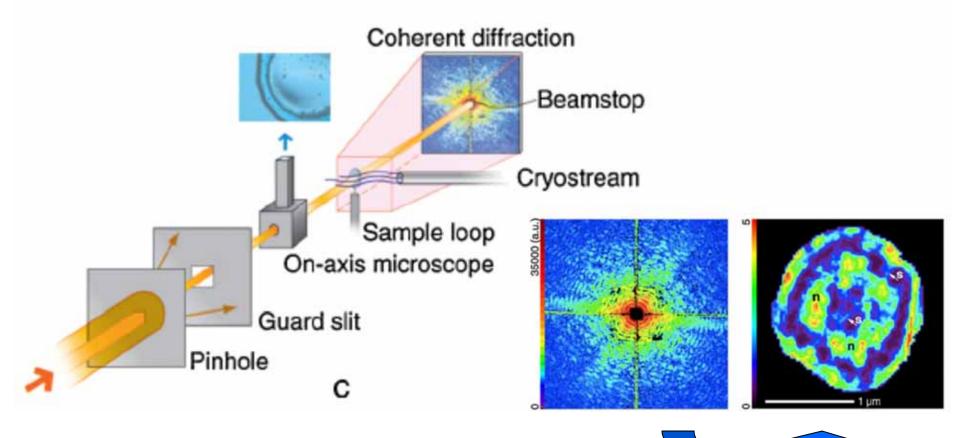
water

XDM of Biological Samples

X-ray Diffraction Microscopy (XDM)

- High penetration power of X-rays
- Biological cells (a few µm) at 10 nm or higher resolution
- Fills the gap between light microscopy (low resolution, ~ 200 nm) and electron microscopy (only thin samples, ~ 0.5 μm)
- Lens-less method
- Image resolution is limited by X-ray diffraction

Experimental Setup



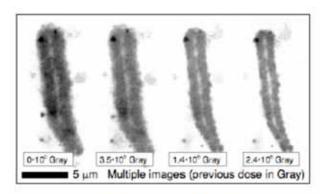
Lima et al. (2009), PRL 103, 198102

Oversampling phasing method ⁴

Sample Preparation

Dehydration

- Chemically fixed or freeze-dried
- Structure degradation/distortion during dehydration process
- Mass loss at 10⁶ Gray (Gy)

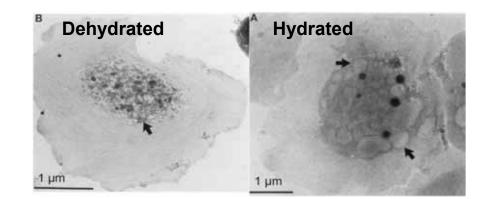


William et al, Journal of Microscopy, 170, pp 155-165 (1993)

- Mass loss at 10⁶ Gray (Gy).
- To reach 10 nm, radiation dose > 10⁸ Gy

Freezing in hydrated state

- Plunge-freezing at 1 bar or highpressure freezing
- Structure close to native state
- Less radiation damage



OToole et al. (1993), J. Stuct. Bio. 110, 55

Recent Progress

- 1. Miao et al. (2003), PNAS 100, 110-112.
- 2. Shapiro et al. (2005), PNAS 102, 15343-15346.
- 3. Jiang et al. (2008), PRL 100, 038103.
- 4. Song et al. (2008), PRL 101, 158101.
- 5. Nishino et al. (2009), PRL 102, 018101.
- 6. Nelson et al. (2010), PNAS 107, 7235.
- 7. Jiang et al. (2010), PNAS 107, 11234.
- 8. Huang et al. (2009), PRL 103, 198102.
- 9. Lima et al. (2009), PRL 103, 198103.

What should be done during freezing ?

Dehydrated

Frozen hydrated

Cryopreservation Concerns

Crystallization must be suppressed during freezing !!

Otherwise,

- 1. Cell structure damage via
 - Solution effect \rightarrow osmotic shock



- Mechanical force due to expansion, etc.
- 2. Parasitic scattering due to density fluctuation
- → Applying high pressure is helpful to suppress crystallization

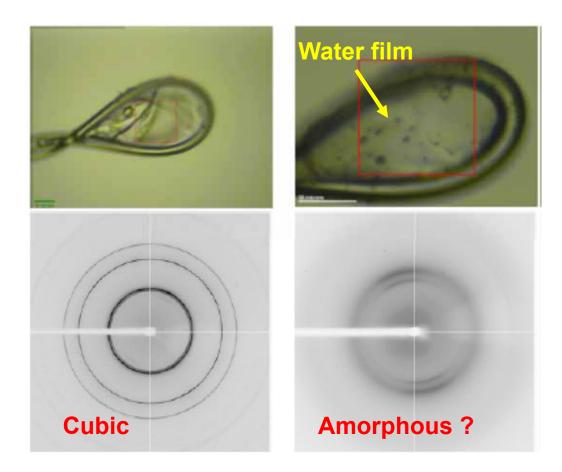
Pressure Effects on Freezing

	1 bar	2 kbar
Crystallization	Fast	Slow
Freezing rate	> 10,000 K/s	~ 100 K/s or slower
Cryo-protectants	High concentration	Low concentration
Ice phase	Amorphous (Low-density)	Amorphous (High-density)

Freezing target: 10 % glycerol solution film (~ 10 um thick)



Plunge-freezing at 1 bar

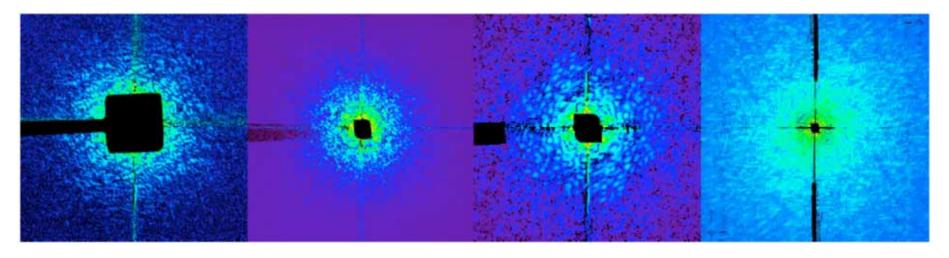


- 1. 10 % glycerol is added
- 2. Plunge-frozen in liquid ethane
- 3. Often cubic ice
- 4. Sometimes diffuse scattering
- →Amorphous or microcubic ice phase ?

Source: Enju Lima (BNL)

Low Success in Reconstruction

Nice speckles but no image reconstruction !!!



Source: Enju Lima (BNL)

May be due to the cubic ice producing parasitic scattering

→ High pressure cryocooling

High Pressure Cryocooling

1. Crystal cryoprotection & diffraction phasing

1) Kim CU, Kapfer R, Gruner SM (2005), Acta Cryst. D61, 881-890.

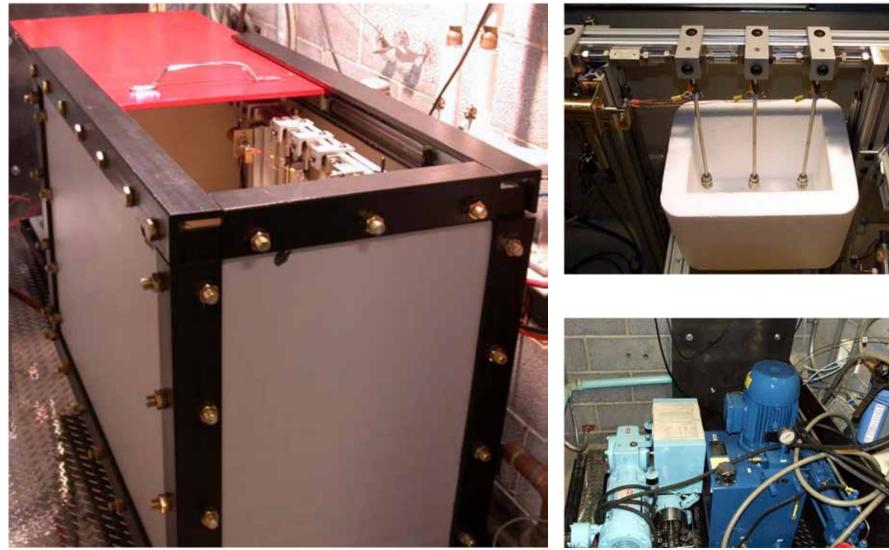
2) Kim CU, Hao Q, Gruner SM (2006), Acta Cryst. D62, 687-694.

- 3) Kim CU, Hao Q, Gruner SM (2007), Acta Cryst. D63, 653-659.
- 4) Kim CU, Chen Y-F, Tate MW, Gruner SM (2008), J. Appl. Cryst. 41, 1-7.

2. Scientific studies

- 1) Albright RA, Ibar JL, Kim CU, Gruner SM, Morais-Cabral JH (2006), *Cell*. **126**, 1147-1159.
- 2) Barstow B, Ando N, Kim CU, Gruner SM (2008), *Proc. Natl. Acad. Sci.*, **105**, 13362-13363.
- 3) Barstow B, Ando N, Kim CU, Gruner SM (2009), Biophys J., 97, 1719 -1727.
- 4) Domsic JF, Avvaru BS, Kim CU, Gruner SM, Agbandje-McKenna M, Silverman DN, McKenna R (2008), *J Biol Chem*, **283**, 30766-30771.
- 5) Kim, CU, Barstow, B, Tate, MW, Gruner, SM (2009), *Proc. Natl. Acad. Sci.*, **106**, 4596-4600.

Apparatus at Gruner Lab

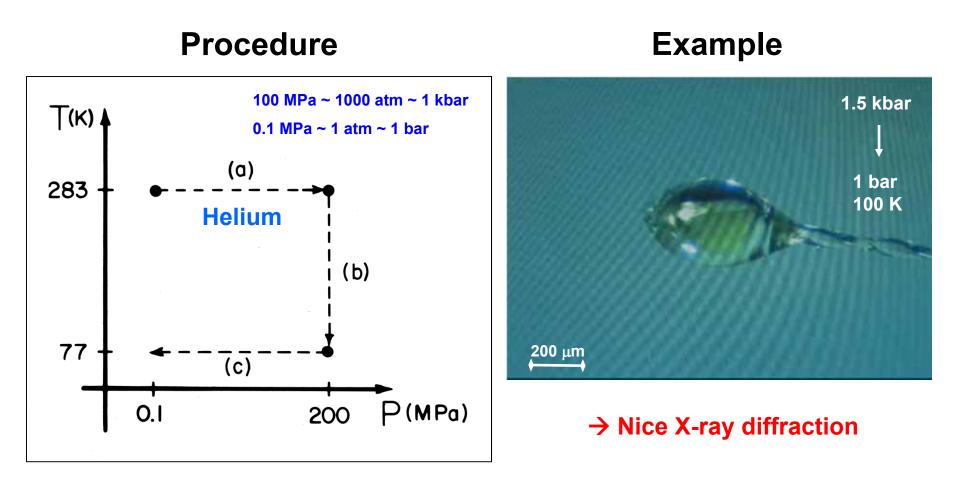


Apparatus at CHESS



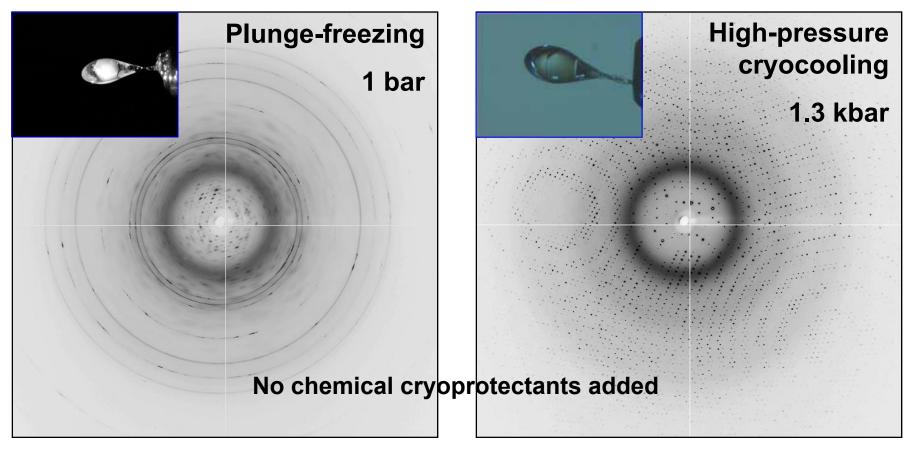
- 1. Double protection with $\frac{1}{2}$ " steel plates.
- 2. All high pressure lines are enclosed.
- 3. Weight ~ 3000 lbs.

High Pressure Cryocooling



Chae Un Kim, Raphael Kapfer & Sol M. Gruner (2005). Acta Cryst. D61, 881-890.

Glucose Isomerase



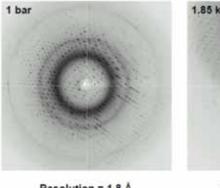
Resolution = 5.0 Å Mosaicity = N/A Resol. = 1.1 Å (1.3 Å for 3 crystals)Mos. = $0.39^{\circ} (0.48^{\circ} \text{ for 3 crystals})$

15

Chae Un Kim, Raphael Kapfer & Sol M. Gruner (2005). Acta Cryst. D61, 881-890.

Examples

Thaumatin

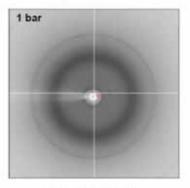


1.85 kbar

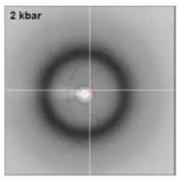
4

Resolution = 1.8 Å Resolution = 1.15 Å
Mosaicity = 1.29" Mosaicity = 0.11"
Chae Un Kim, Rachael Kapler & Sol M. Gruner (2005). Acts Cryst. Del. 881-890.

Pyk2– Schlessinger Group

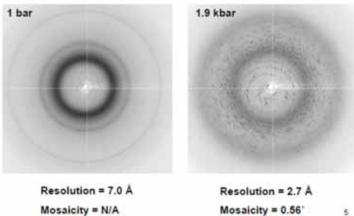


Resolution = 5.0 Å Mosaicity = N/A



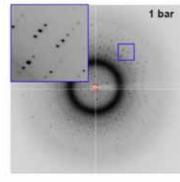
Resolution = 3.0 Å Mosaicity = ~ 0.8' 14

L Amino-acid Oxidase



Chae Un Kim, Raphael Kapfer & Sol M. Gruner (2005). Acta Cryst. D41, 881-890.

T4 lysozyme



1.5 kbar

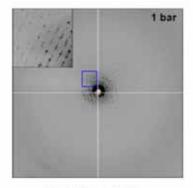
Resolution = 1.7 Å Mosaicity = 0.49

Resolution = 1.20 Å Mosaicity = 0.24'

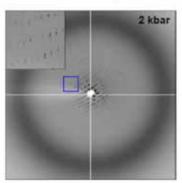
6

Examples

Ribosome - Noller Group



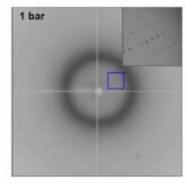
Resolution = 7.0 Å Mosaicity = N/A



12

Resolution = 4.0 Å Mosaicity = 0.74'

SpaD - Baker Group

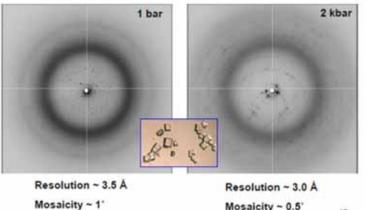


2 kbar

Resolution = 6.0 ~ 2.75 Å 1 indexable data out of > 100 xtals

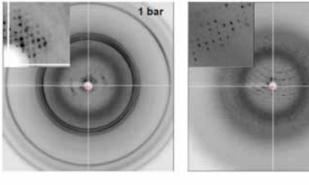
Resolution = 2.2 ~ 2.1 Å 3 indexable data out of 30 xtals

Kv1.2 K+ Ion Channel



Mosaicity ~ 0.5' 10

Cytochrome C Oxidase

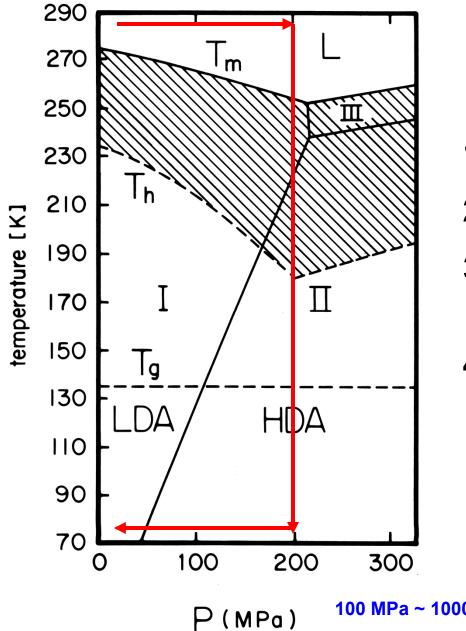


Resolution ~ 8 Å

Resolution ~ 3 Å 11

7

2 kbar



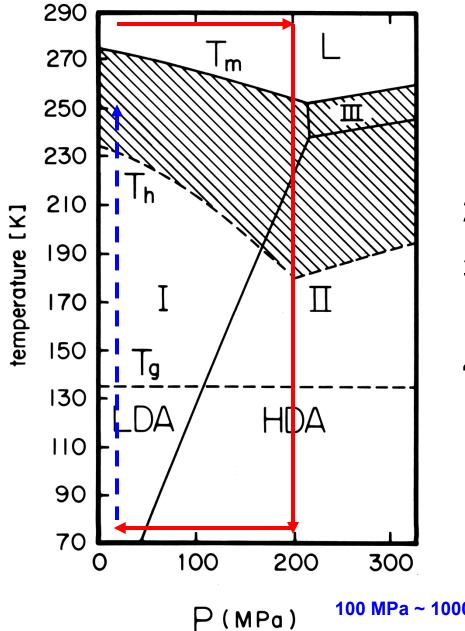
Phase Diagram of H₂O

- 1. Liquid water: L
- 2. Crystalline ice: I, II, III
- 3. Hatched region: supercooled liquid water
- 4. Amorphous ice: Low density amorphous (LDA) High density amorphous (HDA)

Note: Phase boundary is an estimate

100 MPa ~ 1000 atm ~ 1 kbar

Chae Un Kim, Raphael Kapfer & Sol M. Gruner (2005). Acta Cryst. D61, 881-890.



Phase Diagram of H₂O

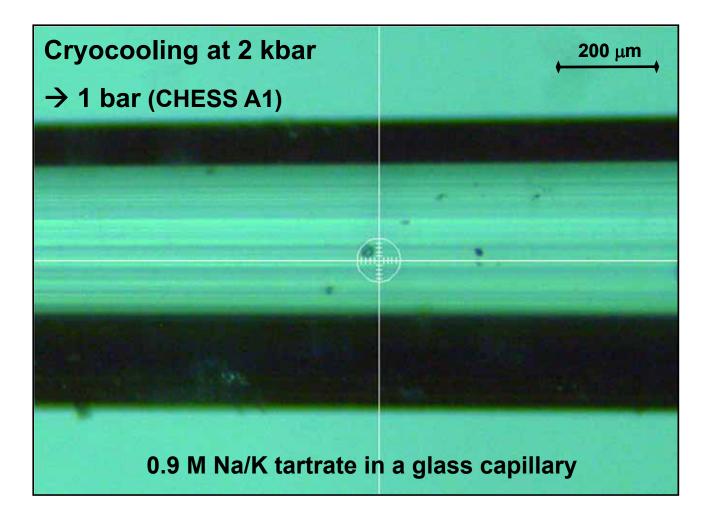
- 1. Liquid water: L
- 2. Crystalline ice: I, II, III
- 3. Hatched region: supercooled liquid water
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Note: Phase boundary is an estimate

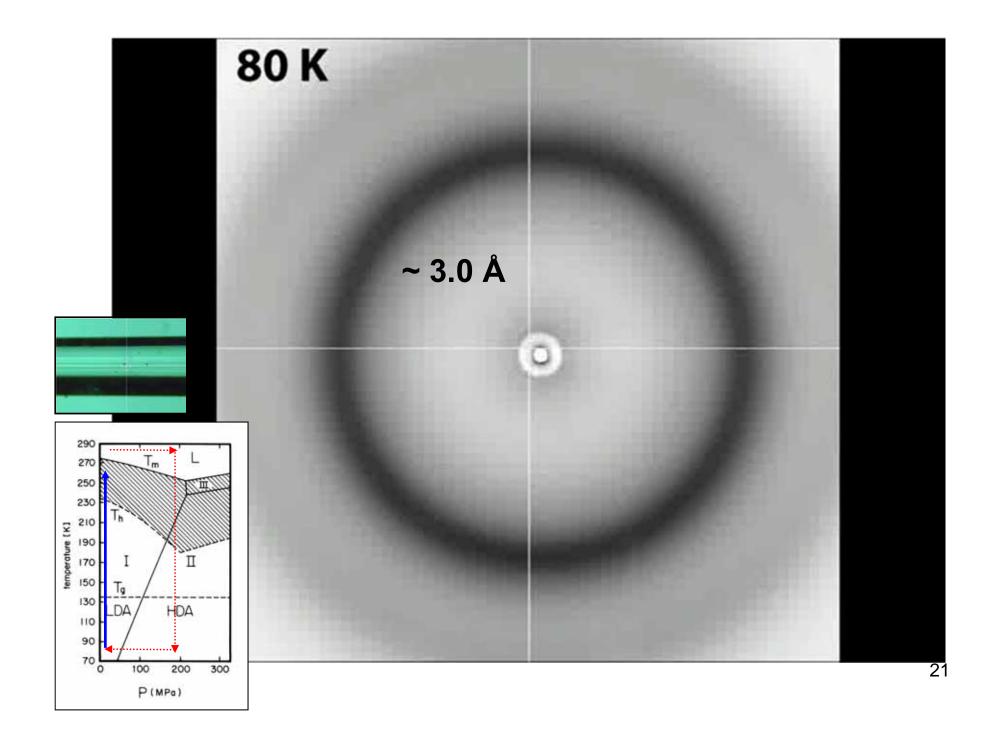
100 MPa ~ 1000 atm ~ 1 kbar

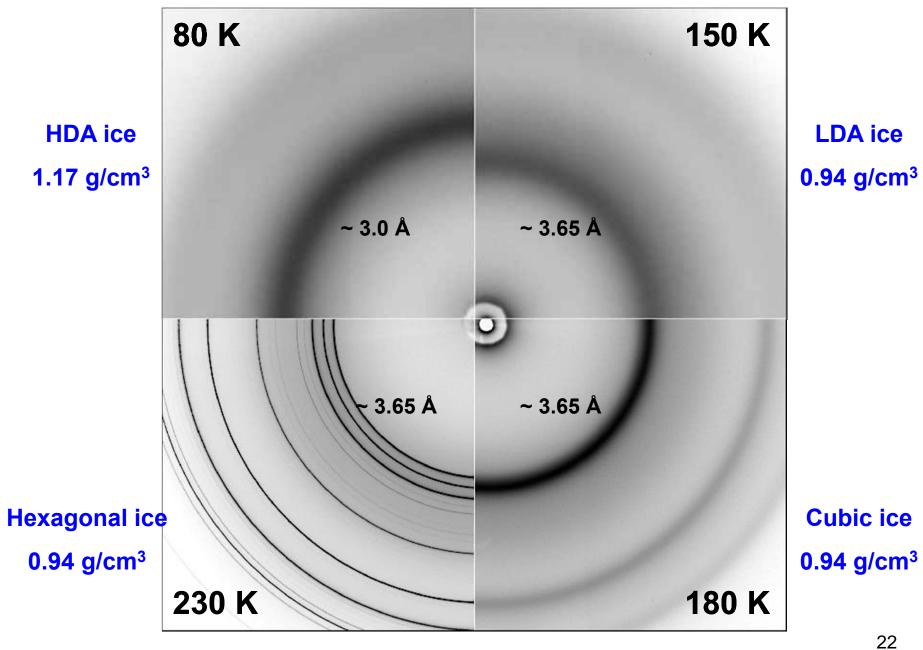
Chae Un Kim, Raphael Kapfer & Sol M. Gruner (2005). Acta Cryst. D61, 881-890.

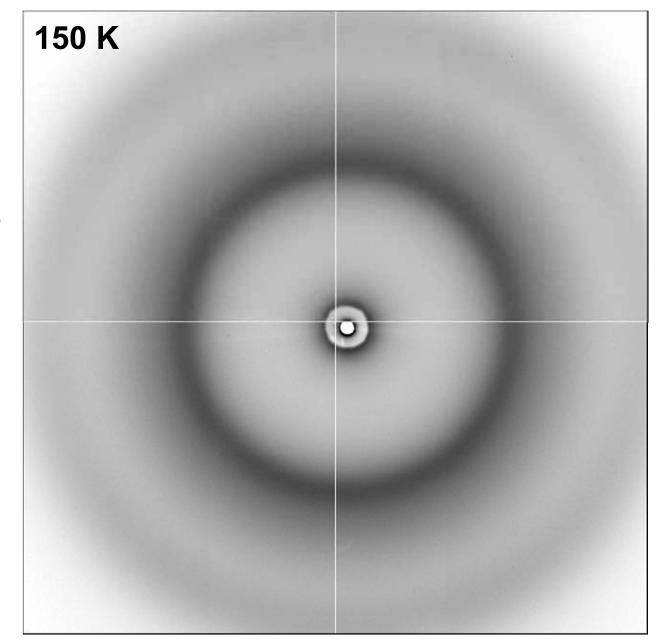
Bulk Solution

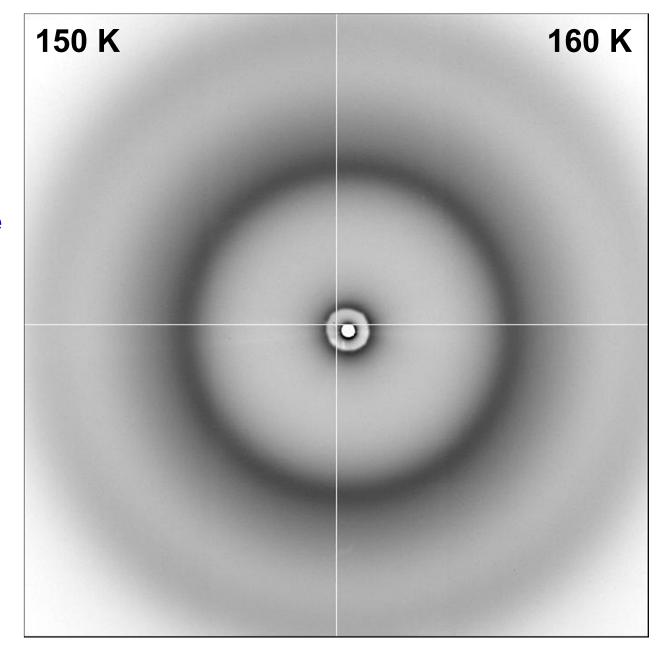


Glass capillary : Inner diameter of 500 μ m, wall thickness of ~ 10 μ m $_{20}$

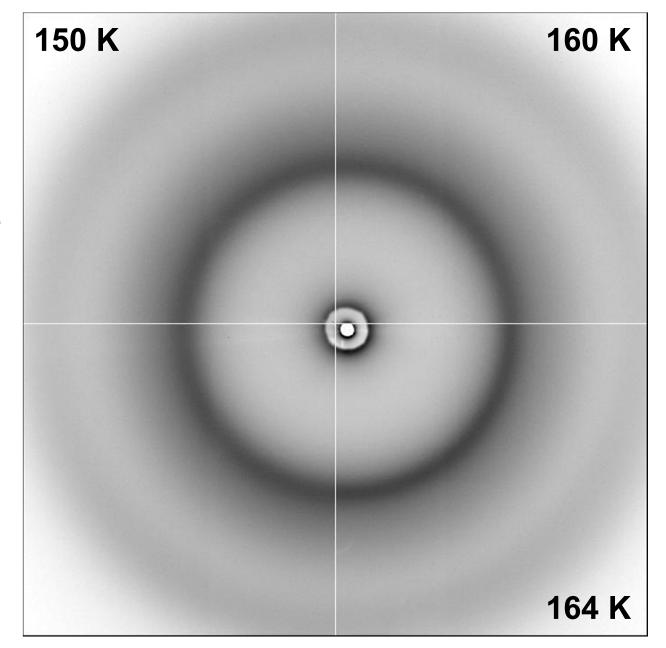


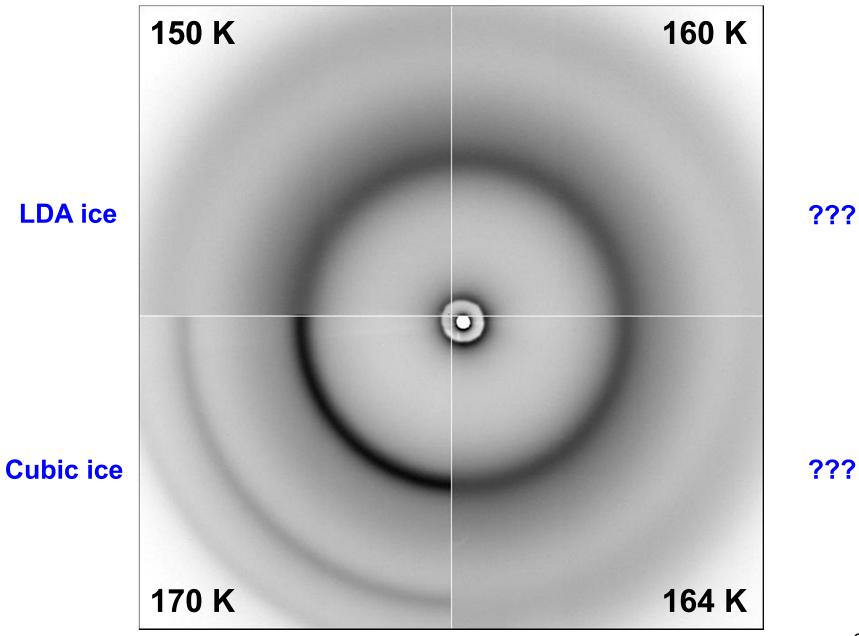


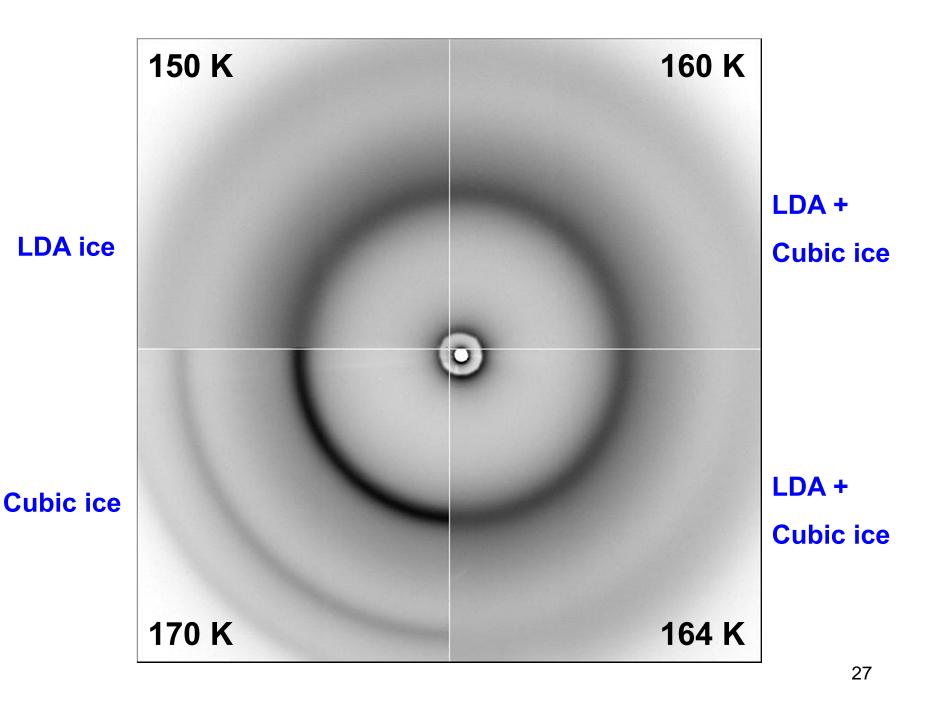


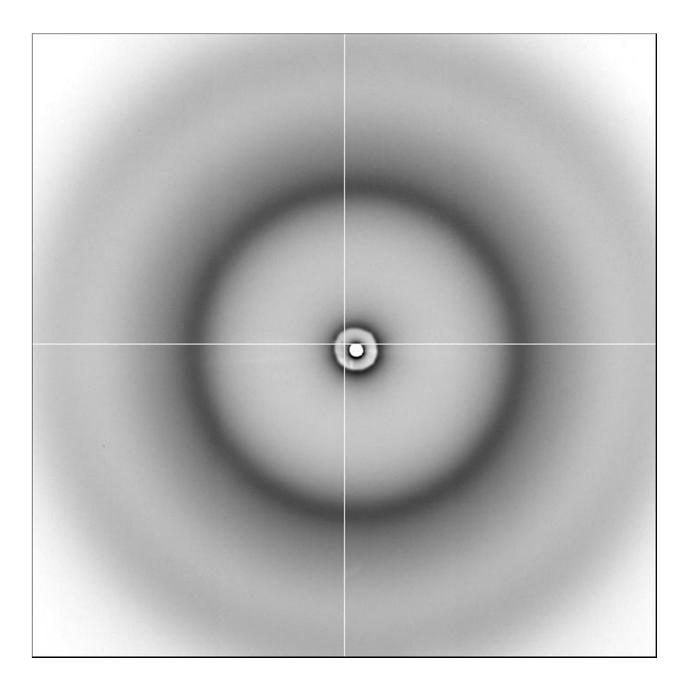




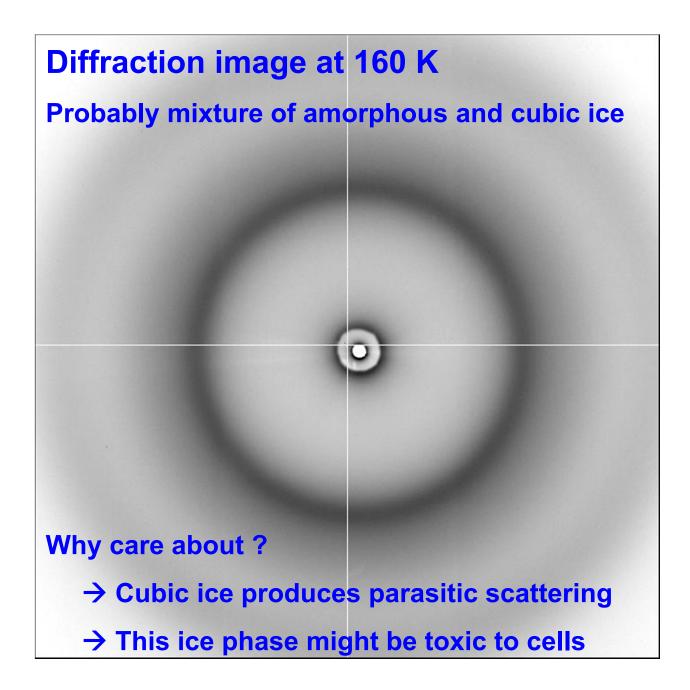




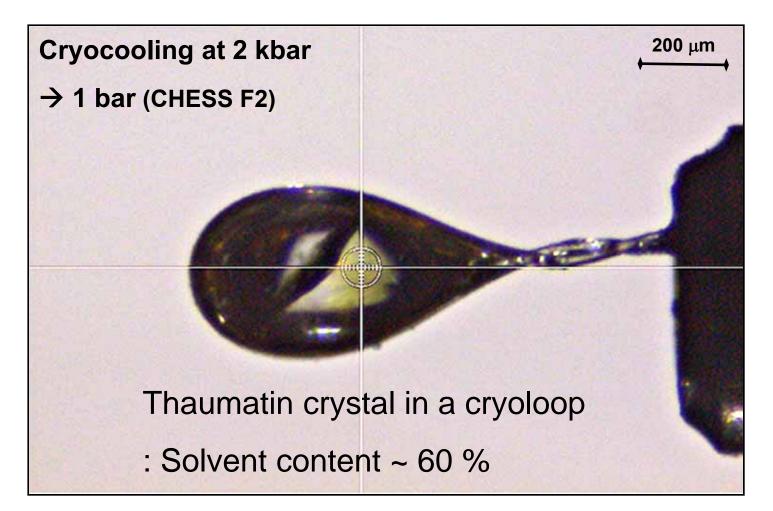




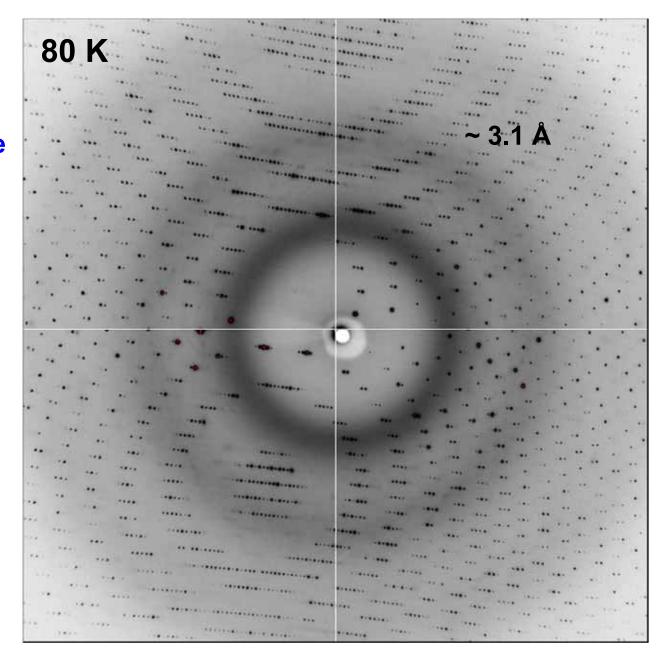
LDA ???



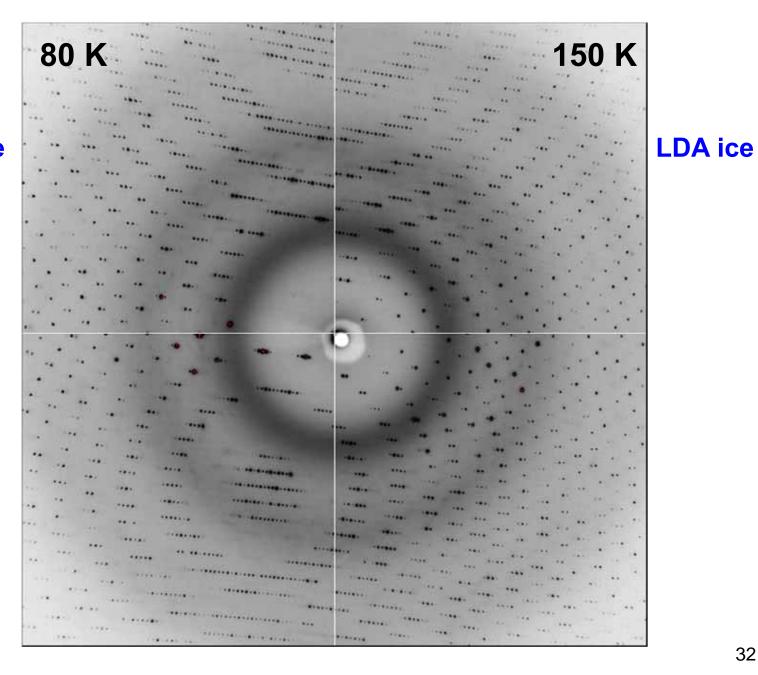
Protein Crystal



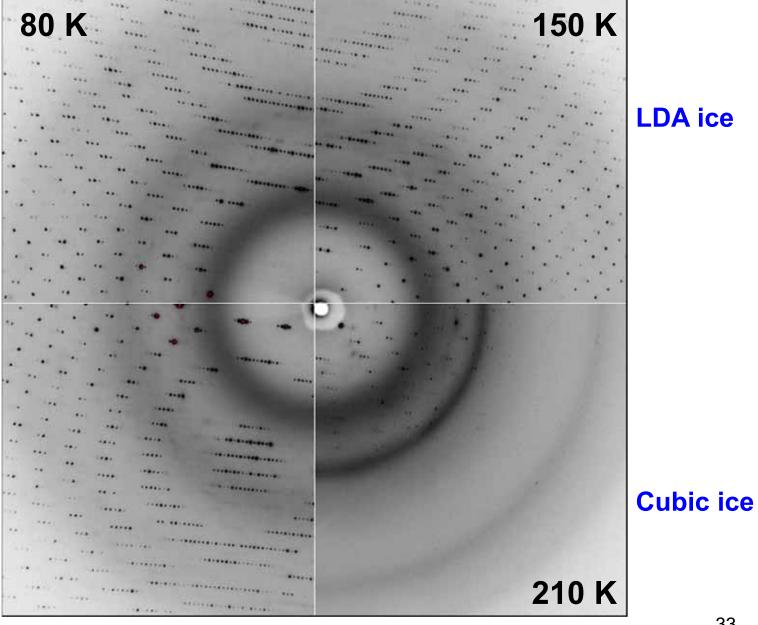
HDA ice

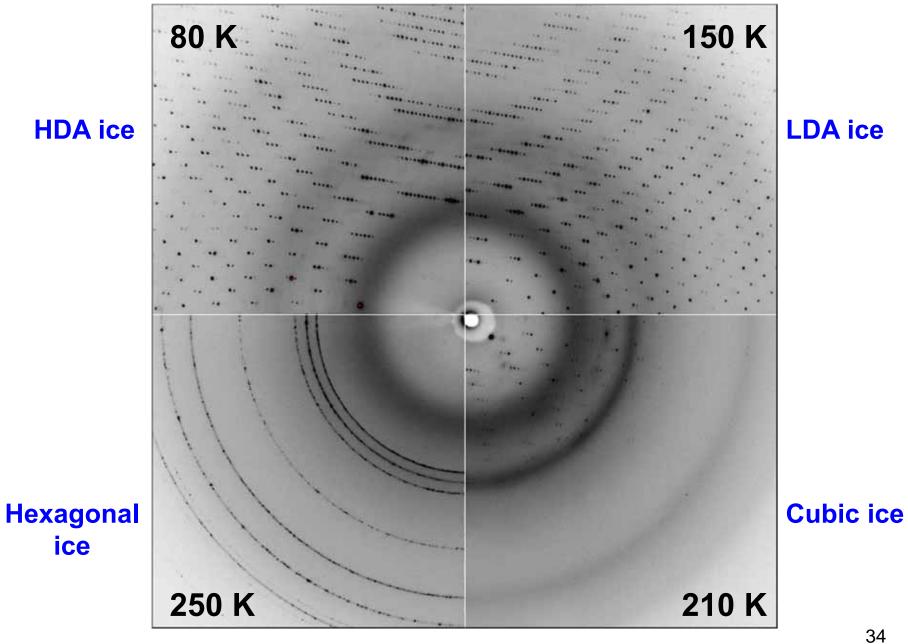


HDA ice



HDA ice





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LDA ice

Cubic ice

160 K 150 K 170 K 180 K

LDA + Cubic ice

LDA + Cubic ice

Implications

- 1. Wide angle X-ray diffraction does not recognize the cubic ice formation until it is quite much developing.
- 2. Plunge-freezing at ambient pressure may end up with the mixture of amorphous and cubic ice phases.
 - \rightarrow Low convergence in image reconstruction

On the other hand,

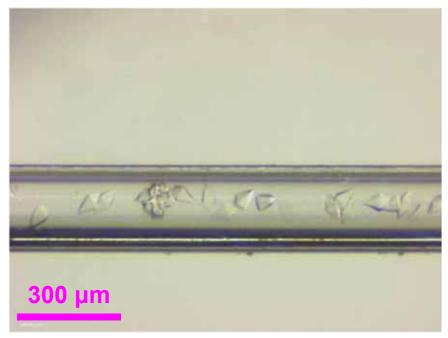
High pressure cryocooling starts with high-density amorphous phase, ensuring amorphous phase.

Methods for Crystal Hydration

Oil-coating



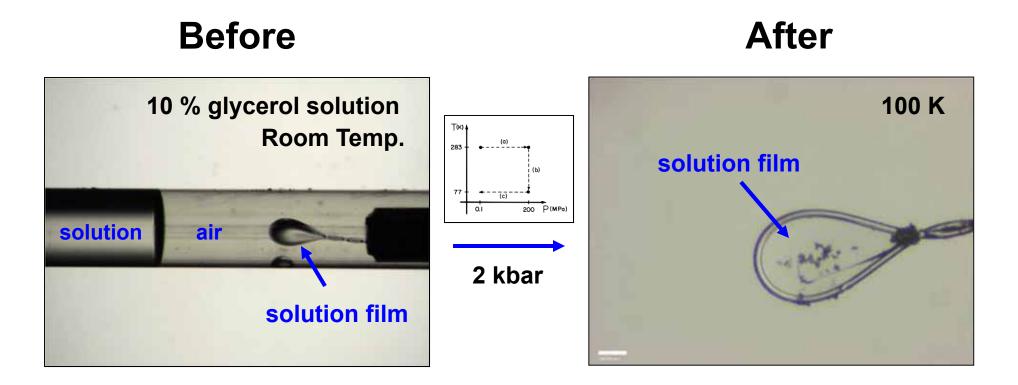
Capillary Hydration



Application for XDM

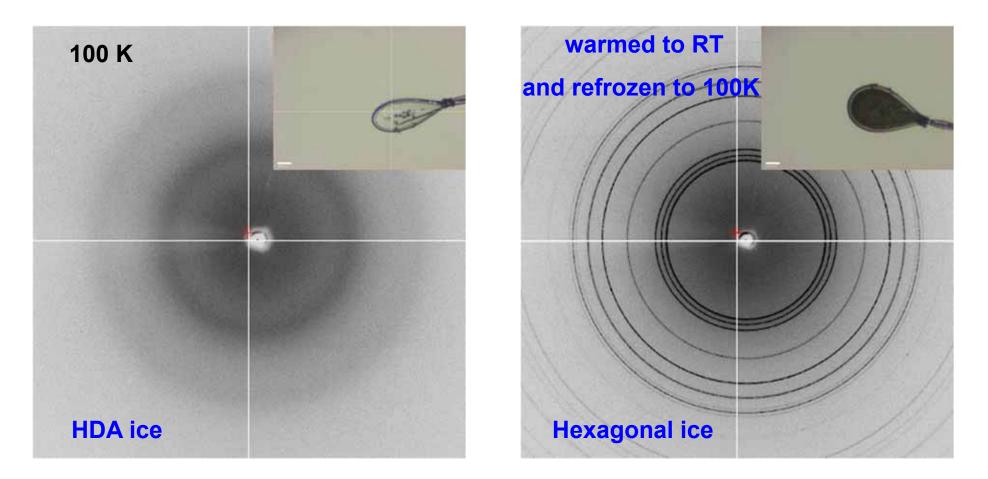
After **Before** 100 K 10 % glycerol solution Т(к) Room Temp. 283 77 (c) 200 P(MPa) 0.1 solution air 2 kbar air solution film solution film

Application for XDM



Background scattering from oil and capillary can be removed !!

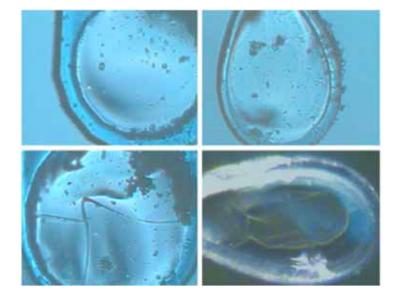
Application for XDM



Collaboration w/ Enju Lima (BNL)

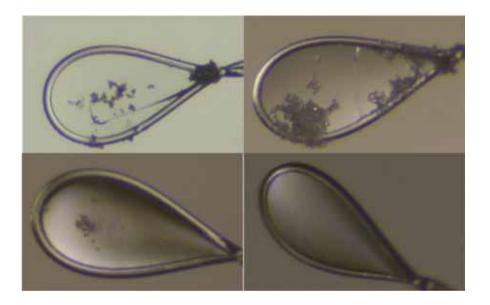
Samples under Optical Microscope

Plunge-freezing at 1bar



Source: Enju Lima (BNL)

High-pressure Cryocooling



Source: Chae Un Kim (CHESS)

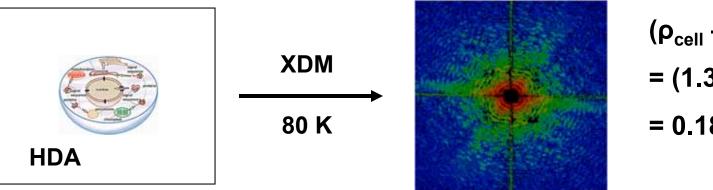
First Trial in ESRF

First Trial in ESRF



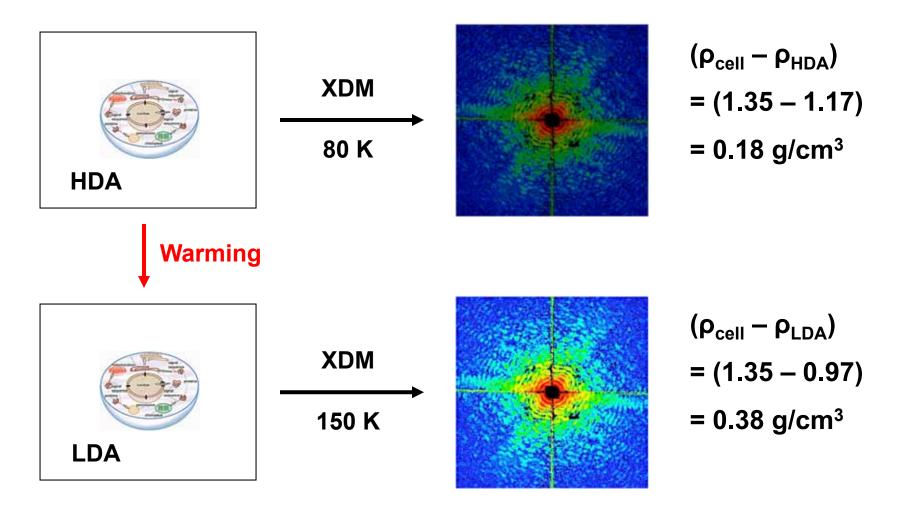
Collaboration w/ Enju Lima (BNL)

Data Collection



 $(\rho_{cell} - \rho_{HDA})$ = (1.35 - 1.17) = 0.18 g/cm³

Data Collection



 $I_{gain} = (0.38/0.18)^2 = 4.46 !!$

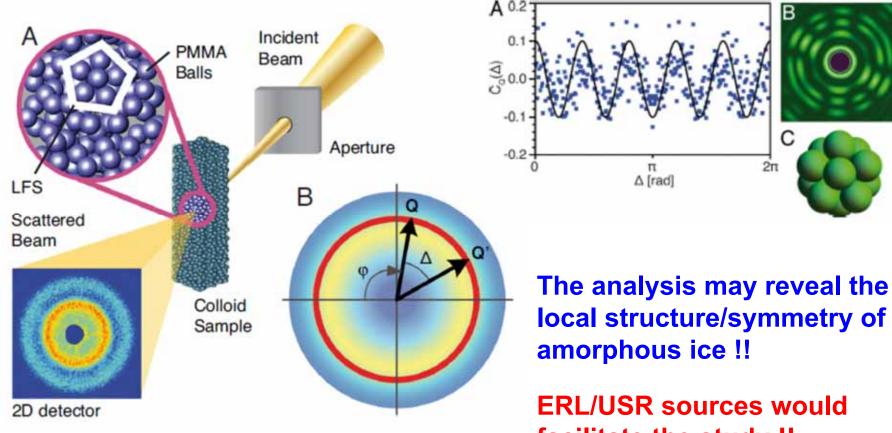
48

Remaining Questions

- 1. Is high-density amorphous (HDA) ice homogeneous in nm scale ?
- 2. Is low-density amorphous (LDA) ice homogeneous in nm scale ?
- 3. Is LDA ice purely amorphous, not micro-cubic phase ?

→ Coherent nano X-ray beam would be helpful !!

X-ray Cross Correlation Analysis



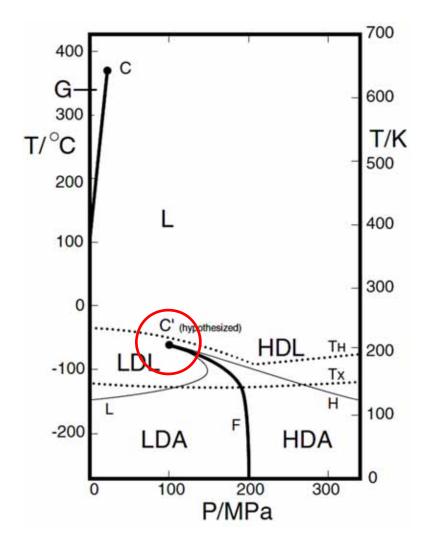
local structure/symmetry of amorphous ice !!

C

ERL/USR sources would facilitate the study !!

Wochner et al. (2009), PNAS 106, 11511

Water and 2nd critical point



- 1. Water shows mysterious thermodynamic properties when supercooled
- 2. 2nd critical point was proposed in low T & high P
- 3. 2nd critical point involves HDA/LDA and their liquid counter-parts
- 4. No structures of HDA/HDL

Mishima & Stanley (1998), *Nature* **396**, 329.

Summary

1. Amorphous ice is required for freezing biological samples in hydrated state.

2. High pressure cryocooling has potential for XDM of biological samples.

3. ERL/USR sources would be helpful to study water structures during phase transition.

Acknowledgement



www.chess.cornell.edu

Sol M. Gruner (Cornell Univ.) Enju Lima (BNL)



The End