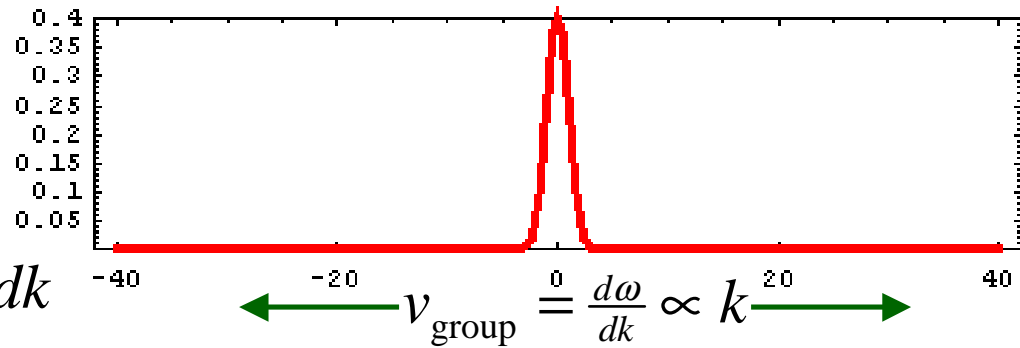


# Dispersion of a free particle packet states

04/04/2005

$$\Psi(x,0) = \frac{1}{\sqrt{\sqrt{\pi}\sigma}} e^{-\frac{x^2}{2\sigma^2}} = f(x)$$



$$\Psi(x,0) = \frac{1}{\sqrt{\sqrt{\pi}\sigma^{-1}}} \int_{-\infty}^{\infty} e^{-\frac{k^2}{2\sigma^{-2}}} \frac{e^{ikx}}{\sqrt{2\pi}} dk$$

$$\Psi(x,t) = \frac{1}{\sqrt{\sqrt{\pi}\sigma^{-1}}} \int_{-\infty}^{\infty} e^{-\frac{k^2}{2\sigma^{-2}}} \frac{e^{i(kx-\omega t)}}{\sqrt{2\pi}} dk, \quad \omega = \frac{\hbar}{2m} k^2$$

$$= \frac{(2\pi)^{-1}}{\sqrt{\sqrt{\pi}\sigma^{-1}}} e^{-\frac{1}{2} \frac{x^2}{\sigma^2 + i\frac{\hbar}{m}t}} \int_{-\infty}^{\infty} e^{-\frac{1}{2}(\sigma^2 + i\frac{\hbar}{m}t)(k + i\frac{x}{\sigma^2 + i\frac{\hbar}{m}t})^2} dk = \frac{(2\pi)^{-1}}{\sqrt{\sqrt{\pi}\sigma^{-1}}} e^{-\frac{1}{2} \frac{x^2}{\sigma^2 + i\frac{\hbar}{m}t}} \frac{\sqrt{2}}{\sqrt{\sigma^2 + i\frac{\hbar}{m}t}} \sqrt{\pi}$$

$$|\Psi(x,t)|^2 = \frac{1}{\sqrt{\pi} \sqrt{\sigma^2 + (\frac{\hbar}{m\sigma}t)^2}} e^{-\frac{x^2}{\sigma^2 + (\frac{\hbar}{m\sigma}t)^2}}$$

(Gaussian with width)

$$\Delta x = \frac{1}{\sqrt{2}} \sqrt{\sigma^2 + (\frac{\hbar}{\sigma m}t)^2}$$

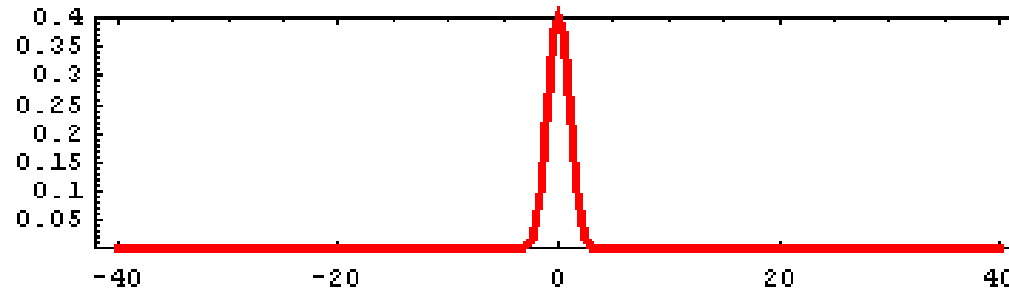
Position-momentum uncertainty relation:

$$\Delta x \geq \frac{\sigma}{\sqrt{2}}, \quad \Delta k = \frac{\sigma^{-1}}{\sqrt{2}} \quad \rightarrow \quad \Delta x \Delta p \geq \frac{\hbar}{2}$$



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Position-velocity uncertainty relation:

$$\Delta x \Delta v \geq \frac{\hbar}{2m} \approx \begin{cases} 1\text{cm} \cdot 1 \frac{\text{m}}{\text{s}} = 1 \overset{\circ}{\text{A}} \cdot 10^8 \frac{\text{m}}{\text{s}} \text{ for an electron} \\ 1\mu\text{m} \cdot 10^{-26} \frac{\text{cm}}{\text{s}} = 1\mu\text{m} \cdot 0.1 \frac{\overset{\circ}{\text{A}}}{14\text{Gyears}} \text{ for a 0.1g pea} \end{cases}$$

**Conclusion:**

In the microscopic world of the hydrogen atom, electron orbits cannot be determined since the uncertainty in velocity is as large as the electron velocity itself.

In the macroscopic world, the uncertainty of quantum mechanics is not recognizable.



## Projection amplitudes and ket vectors for free states

04/04/2005

Gaussian probability distribution at  $t=0$ :  $\Psi(x,0) = \frac{1}{\sqrt{\sqrt{\pi}\sigma}} e^{-\frac{x^2}{2\sigma^2}}$

$$|\Psi\rangle = \sum_{\text{all } x} |x\rangle \langle x|\Psi\rangle = \sum_{\text{all } x} |x\rangle \frac{1}{\sqrt{\sqrt{\pi}\sigma}} e^{-\frac{x^2}{2\sigma^2}} \quad |\Psi\rangle = \sum_{\text{all } k} |k\rangle \frac{1}{\sqrt{\sqrt{\pi}\sigma^{-1}}} e^{-\frac{k^2}{2\sigma^{-2}}}$$

$$\langle x|\Psi\rangle = \sum_{\text{all } k} \langle x|k\rangle \langle k|\Psi\rangle \rightarrow \langle x|k\rangle = \frac{e^{ikx}}{\sqrt{2\pi}}$$

$$\langle k|\Psi\rangle = \sum_{\text{all } x} \langle k|x\rangle \langle x|\Psi\rangle \rightarrow \langle k|x\rangle = \frac{e^{-ikx}}{\sqrt{2\pi}}$$

Satisfies the requirements for projection amplitudes:  $\langle x|k\rangle = \langle k|x\rangle^*$

And  $\Phi_k(x) = \langle x|k\rangle$  is a stationary state of the free Schrödinger equation:

$$-\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \Phi_k(x) = E \Phi_k(x), \quad E = \frac{\hbar^2 k^2}{2m}$$

While these stationary basis states cannot be normalized to 1, any wave function  $\Psi$  is normalized to 1.



## The uncertainty principle and ground-state energies

Area in which the ground state wave function has to be confined  $\Delta x$

Momentum uncertainty in all three dimensions  $\Delta p_x \geq \frac{\hbar}{2} \frac{1}{\Delta x}$

$$E_0 \approx \frac{3}{2m} \left( \frac{\hbar}{2\Delta x} \right)^2$$

Example: Hydrogen atom

$$\left. \begin{aligned} E_0 &= \frac{Ze^2}{8\pi\epsilon_0} \frac{1}{a_0} \\ a_0 &= \frac{4\pi\epsilon_0\hbar^2}{Ze^2m} \end{aligned} \right\} \rightarrow E_0 = \frac{\hbar^2}{2ma_0} \frac{1}{a_0} = \frac{4}{2m} \left( \frac{\hbar}{2a_0} \right)^2$$

This provides a quick procedure of estimating ground state energies.

Example: Hydrogen:  $\hbar \approx 10^{-34}$  Js,  $m_e \approx 10^{-30}$  kg,  $\Delta x \approx 10^{-10}$  m

$$E_0 \approx \frac{3}{2 \cdot 10^{-30} \text{ kg}} \left( \frac{10^{-34} \text{ Js}}{10^{-10} \text{ m}} \right)^2 = 1.5 \cdot 10^{-18} \text{ J} = \frac{1.5}{1.6} 10 \text{ eV}$$

Neutron or proton confined in the nucleus:

$$m_p \approx 10^{-27} \text{ kg}, \quad \Delta x \approx 10^{-14} \text{ m}$$

$$E_0 \approx \frac{3}{2 \cdot 10^{-27} \text{ kg}} \left( \frac{10^{-34} \text{ Js}}{10^{-14} \text{ m}} \right)^2 = 1.5 \cdot 10^{-13} \text{ J} \approx 1 \text{ MeV}$$

