

Nuclei

① Nuclear Structure

- * atomic number Z
- neutron # N
- mass # $A = N + Z$

Z determines electronic structure \rightarrow chemistry

Notation : $\begin{array}{c} A \\ \textcircled{Z} \\ X \end{array}$ sometimes here
sometimes drop \rightarrow chemical element symbol

- * isotopes : same Z (different N)

e.g. $\begin{array}{ll} {}^{12}\text{C} & 98.9\% \\ {}^{13}\text{C} & 1.1\% \\ {}^{14}\text{C} & \text{unstable } 5,730 \text{ yr half-life} \end{array}$ natural abundance

isobars : same A

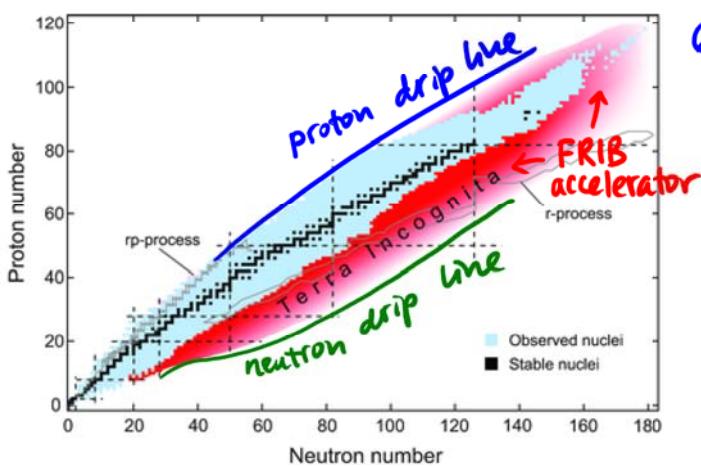
${}^{12}\text{B}$, ${}^{12}\text{C}$

isotones : same N (replace p with n in 'isotope')

${}^{12}\text{B}$, ${}^{13}\text{C}$

isomer : excited nucleus of the same type
(e.g. prior to γ emission)

${}^{99m}\text{Tc}^{(*)}$ $\left. \begin{array}{l} 140\text{keV X-ray} \\ \text{"technetium"} \\ \text{6hr half-life} \end{array} \right\}$ (medical app)



② Nuclear stability due to nuclear force!

$$Z \approx N, \text{ for } Z \leq 20$$

$$N \geq Z, \text{ larger } Z$$

"shell" structure Z or N :
2, 8, 20, 28, 50, 82, 126
"magic numbers"

③ Nucleons

* neutron (n) charge : 0
 mass : 1.008 u ($\frac{1}{12}$ of ^{12}C atomic mass)

* proton (p) charge : $+e$
 mass : 1.007 u

$$\text{Rest energy } E = mc^2, 1\text{ u} = 931.5 \text{ MeV}/c^2$$

* Both p and n size : $\sim 1 \text{ fm}$ ($1 \text{ fm} = 10^{-15} \text{ m}$, "fermi")

Q: can a nucleus contain e^- ?

(Rutherford: there maybe $(A-Z)$ neutral proton-electron pairs in nuclei)

No. confinement to 10^{-14} m gives
 too high KE, $> 10 \text{ MeV}$
 measured β -decay ($= e^-$) $\sim 1 \text{ MeV}$

No. magnetic moment of $e^- \gg$ nuclear moment
 $\mu_n = \frac{e\hbar}{2m_p}$ "nuclear magneton"
 ~ 2000 smaller than μ_B

* nuclear spin I : same for p & n, $I = \frac{1}{2}$
 \downarrow
 nuclear g-factor

* magnetic moment : $\mu_I = g_I m_I \mu_n$
 (p) $2.79 \mu_n$
 (n) $-1.91 \mu_n$
 \uparrow
 spin projection
 quantum number
 $-I \leq m_I \leq I$

* electric dipole moment ($d = ql$)

Standard Model prediction : $|d_{nl}| \sim |d_{pl}| \sim 10^{-32} \text{ e.cm}$

SUSY prediction : $|d_{nl}| \sim 10^{-28}$ to 10^{-25} e.cm

best experiment so far : $|d_{nl}| < 10^{-26} \text{ e.cm}$

Effort to measure $|d_{pl}|$ to $< 10^{-28} \text{ e.cm}$ level
 (proton electrostatic ring accelerator)

④ Nuclei size

Initial data from scattering



$$\text{closest approach : } \frac{1}{2} m v_d^2 = \frac{(2e)(Ze)}{4\pi\epsilon_0 r}$$

Generally:

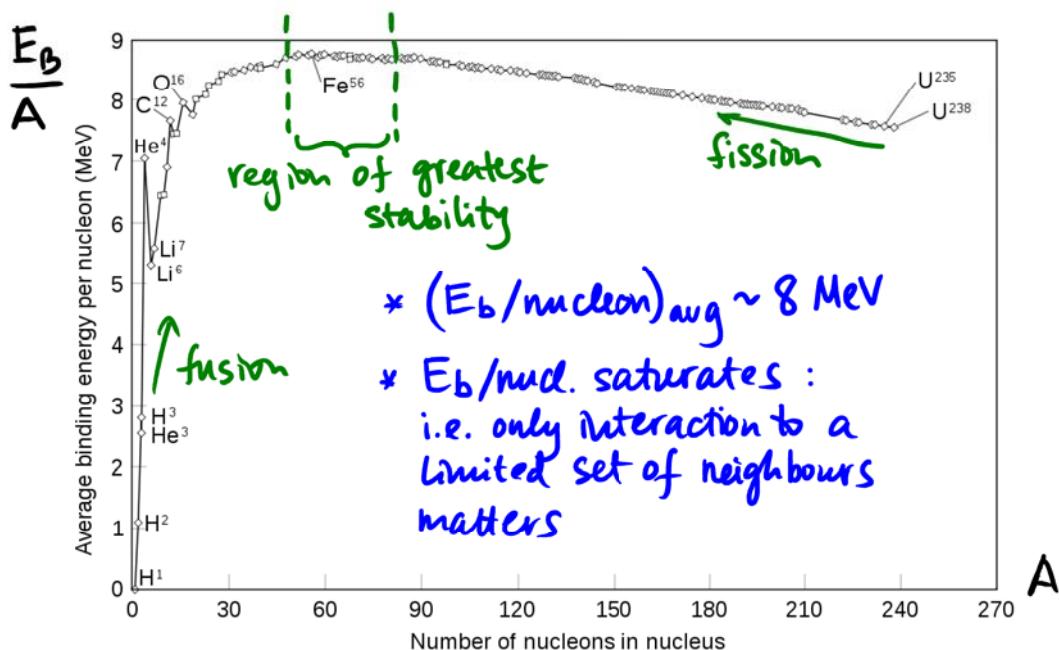
$$r \approx r_0 A^{1/3}, \text{ with } r_0 \approx 1.2 - 1.5 \text{ fm}$$

compare to atoms, which roughly have same size
⇒ different force law

⑤ Binding energy = mass deficit

$$E_b = (Z M(H) + N M_n - M_A) c^2 = \Delta m c^2$$

mass of a nucleus is always less than the sum individual p's and n's that it consists of.



⑥ Nuclear force

- * attractive
- * short range (\sim fm)
- * depends on spin
- * does not depend on charge } from scattering experiments
- * actual force is a sum of Coulomb repulsion (p-p) and nuclear force

E.g. Continuous Electron Beam Accelerator Facility

(CEBAF): e^- probe @ 6 GeV smashing nuclei

$$\lambda_{\text{deBroglie}} = \frac{hc}{pc} = \frac{1.24 \text{ keV-nm}}{6 \text{ GeV}} \sim 0.2 \text{ fm}$$

now being upgraded to 12 GeV, $\lambda \sim 0.1 \text{ fm}$
can see quark structure of nucleons

- * nuclear force is mediated by exchange of virtual particles (more later)

Q: where does the energy come from to create virtual particles?

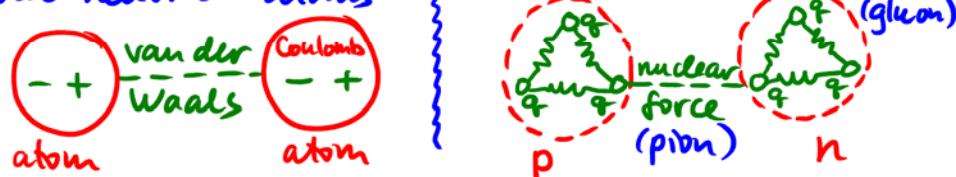
A: nowhere, uncertainty principle allows for large fluctuations of energy if $\Delta t \rightarrow 0$

$$\Delta t \lesssim \frac{\hbar}{mc^2} \text{ virtual particle mass}$$

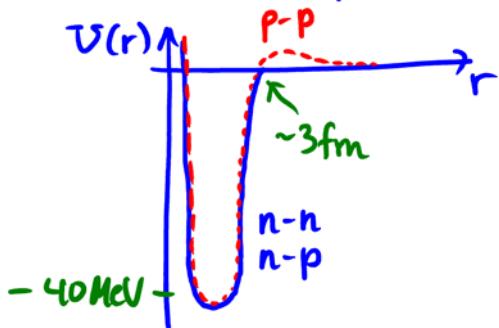
$$\text{range} = \Delta t c \lesssim \frac{\hbar}{mc} \approx 2 \text{ fm}, \Rightarrow mc^2 \sim 100 \text{ MeV}$$

π meson, Yukawa (pion)

- * nuclear force between nucleons: "residual"
similar to Van der Waals force between two neutral atoms



Ex. nuclear potential



- * minimum in potential
~ 40 to 50 MeV
- * drops to zero @ ~3 fm
- * strongly repulsive $r < 0.4 \text{ fm}$
- * p-p positive potential barrier
~ 1 MeV @ 4 fm

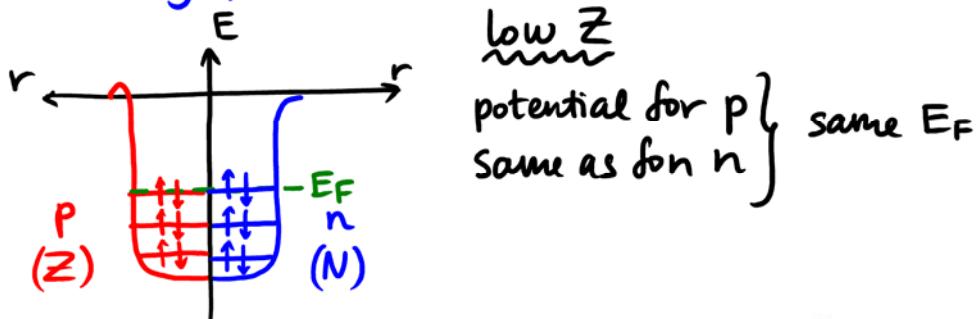
⑦ Nuclear models

- * True theory: must refer to quarks, force b/w them = "strong interaction"
- * Quantum Chromodynamics \Rightarrow all nuclear structure can be explained in principle
color; gluons carry "color"

Here, phenomenological models only

a) Fermi gas model

independant fermions moving within net binding potential



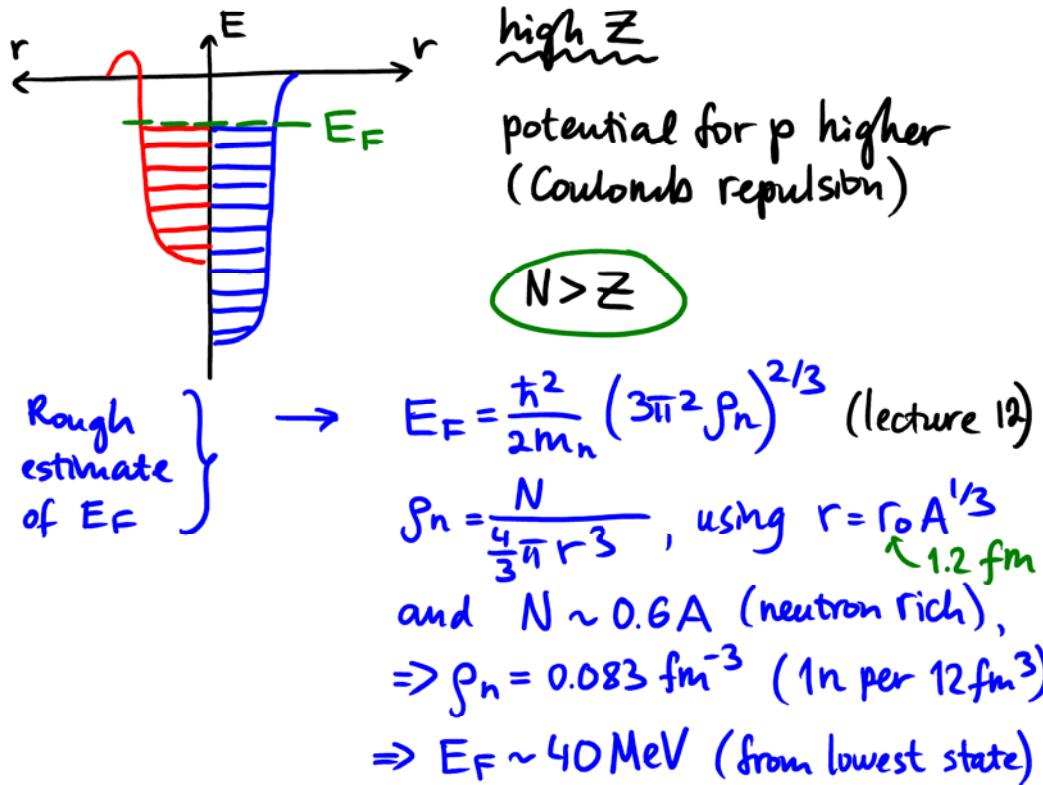
Q: What if E_F is different for p's and n's?

$$(m_n - m_p)c^2 = 1.3 \text{ MeV} \text{ (small)}$$

if free

A: In this case ($N \neq Z$), $p \leftrightarrow n$ to level $(E_F)_{n,p}$ out.

Result: $N \approx Z$ explains low Z data

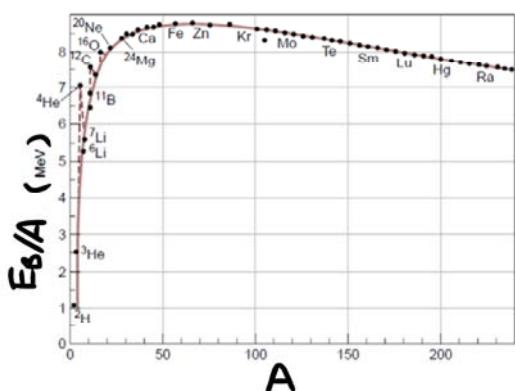


b) Liquid drop model (phenomenological)

$$E_b = C_1 A - C_2 A^{2/3} - C_3 \frac{Z(Z-1)}{A^{1/3}} - C_4 \frac{(N-Z)^2}{A}$$

↑ ↑ ↑ ↑
 volume term surface term Coulomb symmetry of
 (nuclear force (c.f. surface repulsion p - n favoured
 saturation) tension)
 { drop of liquid analogy

$$E_b \text{ in MeV : } C_1 = 15.7, C_2 = 17.8, C_3 = 0.71, C_4 = 23.6$$



← good fit
No explanation
for magic numbers

(next lecture)

More models: c) shell model ; d) collective model